

In Vitro Demonstration of Apoptosis Mediated Photodynamic Activity and NIR Nucleus Imaging through a Novel Porphyrin

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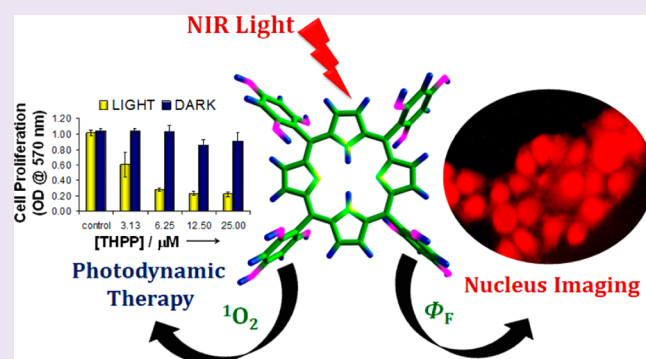
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Supporting Information

ABSTRACT: We synthesized a novel water-soluble porphyrin **THPP** and its metalated derivative **Zn-THPP** having excellent triplet excited state quantum yields and singlet oxygen generation efficiency. When compared to U.S. Food and Drug Administration approved and clinically used sensitizer Photofrin, **THPP** showed *ca.* 2–3-fold higher *in vitro* photodynamic activity in different cell lines under identical conditions. The mechanism of the biological activity of these porphyrin systems has been evaluated through a variety of techniques: 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) assay, comet assay, poly(ADP-ribose)-polymerase (PARP) cleavage, CM-H₂DCFDA assay, DNA fragmentation, flow cytometric analysis, fluorescence, and confocal microscopy, which confirm the apoptotic cell death through predominantly reactive oxygen species (ROS). Moreover, **THPP** showed rapid cellular uptake and are localized in the nucleus of the cells as compared to Hoechst dye and Photofrin, thereby demonstrating its use as an efficient sensitizer in photodynamic therapy and live cell NIR nucleus imaging applications.



Photodynamic therapy (PDT) is an emerging treatment for different types of cancers.^{1–4} It involves the inactivation of living cells by the combined action of light, a photoactive molecule called sensitizer and molecular oxygen.^{5–7} A variety of sensitizers have been examined for their use in PDT. These include porphyrins, chlorins, bacteriochlorins, phthalocyanines, metallophthalocyanines, porphycenes, squaraines, cyanines, rose bengal, methylene blue, aminolevulinic acid, and their derivatives.^{8–11} Among these, porphyrins have attracted much attention as sensitizers due to their higher cellular affinity and very low dark toxicity.^{12–14} For example, the hematoporphyrin derivative (HpD) and its commercial variant, i.e., Photofrin, is being currently used in the treatment of various types of cancers.^{15–17} However, HpD is a mixture of at least nine components, causes cutaneous photosensitivity and immunosuppression, and, more importantly, has only a weak absorption in the red region of the spectrum.^{18,19} The drawbacks associated with the first generation photosensitizer HpD have provided the great impetus to search for new photosensitizers with greater specificity and efficacy.^{20,21} In this context, we designed a novel porphyrin derivative **THPP** and have investigated its photophysical and *in vitro* photobiological

properties when compared to its zinc complex, **Zn-THPP**, and U.S. Food and Drug Administration approved and clinically used sensitizer, Photofrin. Interestingly, **THPP** showed high solubility in buffer and significant quantum yields of triplet excited state and singlet oxygen generation. Uniquely, **THPP** exhibited *ca.* 2–3-fold high photocytotoxicity and localized specifically in the nucleus when compared to Photofrin, indicating thereby its potential as an efficient sensitizer for PDT and NIR imaging applications.

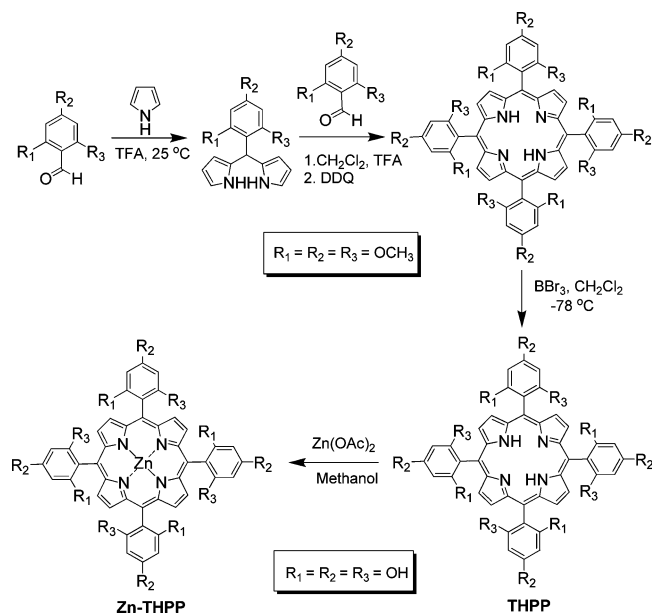
Synthesis of the new porphyrin derivatives **THPP** and **Zn-THPP** was achieved through the modified Lindsey's method^{22,23} (Scheme 1) and were characterized on the basis of elemental analysis and spectral evidence (Supplementary Figures 1 and 2). These derivatives showed high solubility in alcoholic solvents and aqueous medium. The free base **THPP** showed the characteristic absorption of the porphyrin chromophore having a Soret band at 416 nm and four Q-bands at 514, 548, 592, and 648 nm, while its fluorescence

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Scheme 1. Synthesis of Porphyrins THPP and Zn-THPP



spectrum exhibited two emission maxima at 645 and 709 nm (Figure 1a,b). The complex, **Zn-THPP**, similarly exhibited a bathochromically shifted (*ca.* 7 nm) Soret absorption at 423 nm and two Q-bands at 555 and 595 nm, whereas two emission bands were centered at 606 and 656 nm (Supplementary Figure 3). The emission quantum yields (Φ_F) were calculated and are found to be 0.08 ± 0.002 and 0.03 ± 0.001 , for **THPP** and **Zn-THPP**, respectively, in water.

To understand the transient intermediates involved in these systems, we have carried out nanosecond laser flash photolysis studies employing a 532 nm laser pulse (Figure 1c). The

transient absorption spectrum of **THPP** obtained after the laser excitation in methanol showed an absorption maximum at 460 nm with a bleach at 420 nm, where the compound has significant ground state absorption. The lifetime of the transient was calculated from the decay profile at 460 nm (Inset of Figure 1c) and is found to be 12 μs . The transient formed was characterized as a triplet excited state of **THPP** based on the quenching experiments and its negligible formation in the presence of dissolved oxygen. Similar observations were made with **Zn-THPP**. This complex showed the triplet excited state absorption maximum at 450 nm and lifetime of 20 μs in methanol (Supplementary Figure 4). The quantum yields of triplet excited states (Φ_T) were estimated using triplet-triplet energy transfer method to β -carotene and $[\text{Ru}(\text{bpy})_3]^{2+}$ as the standard.^{24,25} The values in methanol are $\Phi_T = 0.66 \pm 0.05$ and 0.94 ± 0.02 , for **THPP** and **Zn-THPP**, respectively.

In addition to favorable photophysical properties, the PDT efficacy of a sensitizer depends on its efficiency of singlet oxygen generation. Therefore, we have determined quantum yields [$\Phi(^1\text{O}_2)$] of singlet oxygen by using 1,3-diphenylisobenzofuran (DPBF) as the singlet oxygen scavenger. To calculate the yields, the sensitizer along with DPBF was irradiated using a xenon lamp with a 515 nm LP filter at different time intervals from 0 to 30 s. The decrease in the absorbance of DPBF was monitored at 410 nm and compared with those observed with the standard, tetra(*meta*-hydroxyphenyl)porphyrin (*m*-THPP) under identical conditions.^{26,27} From the slope of the graph obtained by plotting the change in absorbance of DPBF against the time of irradiation (Figure 1d), we calculated singlet oxygen quantum yields. These values are $\Phi(^1\text{O}_2) = 0.59 \pm 0.03$ for **THPP** and 0.92 ± 0.02 for **Zn-THPP** in methanol.

To investigate *in vitro* photodynamic activity (PDT) of these derivatives, we employed MTT assay in three different cancer

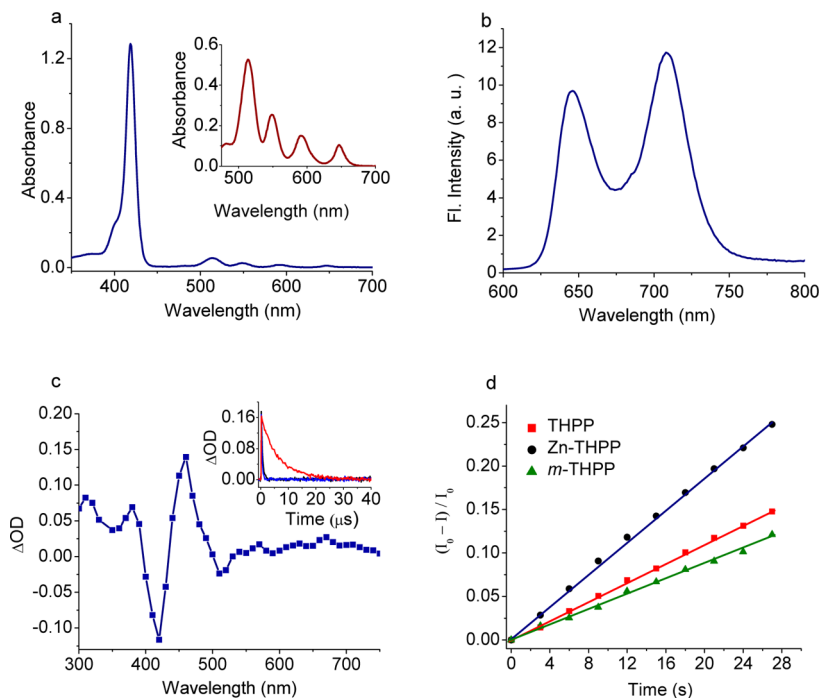


Figure 1. (a) UV-vis absorption and (b) fluorescence (λ_{exc} , 430 nm) spectra of **THPP** (4 μM) in water. (c) Triplet transient absorption spectrum with decay at 460 nm (inset) of **THPP** (20 μM) in methanol and (d) plot of changes in absorption of 1,3-diphenylisobenzofuran with **THPP**, **Zn-THPP**, and *m*-THPP (515 nm LP) versus irradiation time in methanol.

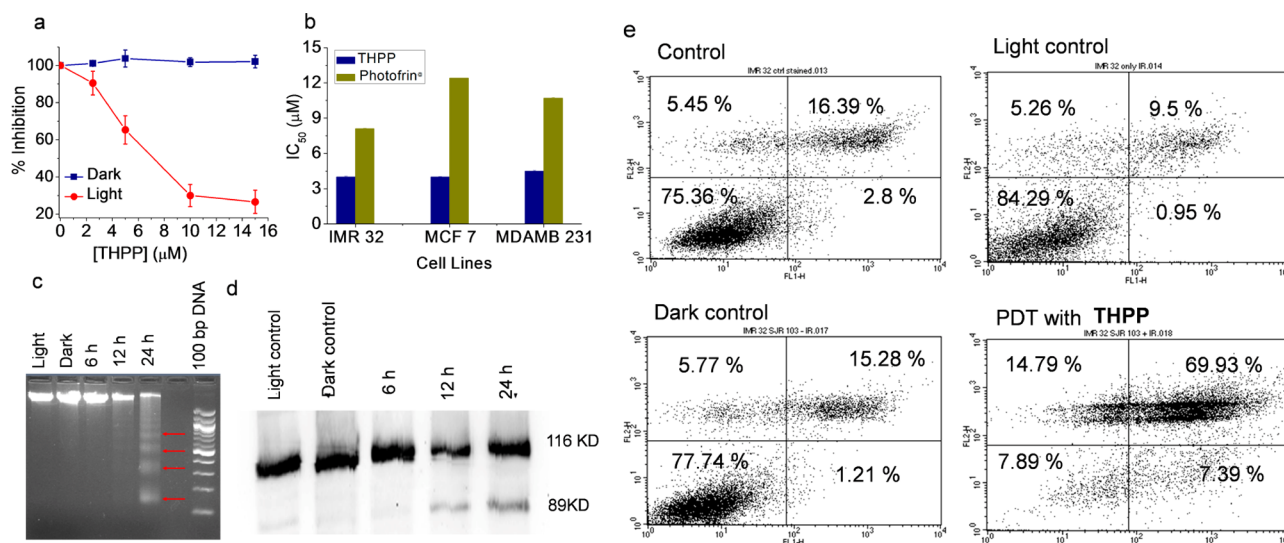


Figure 2. (a) Cytotoxicity of THPP in dark (■) and light (●) in IMR 32 cell lines. (b) Comparison of IC₅₀ values of THPP and Photofrin in IMR 32, MDA MB 231, and MCF 7 cell lines. (c) Fragmentation pattern of genomic DNA after PDT with THPP ($\lambda_{irr} > 600$ nm). (d) Western blot showing poly(ADP-ribose)polymerase cleavage in SCC 131 cells treated with THPP. (e) Flow cytometric analysis of IMR 32 cells after PDT with THPP. The lower left quadrant of each panel shows viable cells, negative for Annexin V-FITC/PI, whereas upper right quadrants represent the apoptotic cells, positive for Annexin V-FITC/PI.

cell lines; IMR 32 (human neuroblastoma), MCF 7 (human breast cancer), and MDA MB 231 (aggressive human breast cancer). The light source used was from a commercially available equipment, Waldman PDT 1200L (600–720 nm, 200J cm⁻², and 50 mW cm⁻²).²⁸ The IC₅₀ value, which is a measure of cytotoxic activity, was calculated for all three cell lines (Figure 2). The PDT activity of THPP and Zn-THPP was compared with the commercial drug, Photofrin, under identical conditions to estimate the efficacy. The percentage cell viability of IMR 32 cells in the presence of THPP under different conditions is shown in Figure 2a. THPP, interestingly, showed high photocytotoxicity and exhibited IC₅₀ values of 4, 4, and 4.5 μM, respectively, in IMR 32, MCF 7, and MDA MB 231 cell lines. As expected, we observed negligible cytotoxicity when these cell lines were treated with THPP in the dark (Supplementary Figures 5 and 6). In contrast, Photofrin exhibited relatively higher IC₅₀ values of 8.12, 12.4, and 10.7 μM, respectively, for these three cell lines (Figure 2b). The complex Zn-THPP, which generated singlet oxygen in near quantitative yields ($\Phi(^1O_2) = 0.92 \pm 0.02$), showed much less photocytotoxicity (IC₅₀ value of 18 μM) in MDA MB 231 cells. This observation clearly indicates that the photodynamic activity of a sensitizer in cells not only depends on singlet oxygen yields but also on other parameters such as cellular uptake and retention. The comparison of IC₅₀ values indicates that THPP exhibits high *in vitro* PDT activity of ca. 2–3-fold higher than that of Photofrin in all the tested cell lines.

To understand the mechanism of biological activity, we analyzed the genomic DNA fragmentation through DNA ladder assay, poly(ADP-ribose)polymerase (PARP) cleavage,²⁹ and comet assay.^{30–32} Upon treatment with THPP followed by irradiation, we observed DNA ladder in oral cancer cells (SCC 131) (Figure 2c), which was prominent after 24 h. However, no such DNA ladder was observed when the cells were either treated with THPP in dark or irradiated without THPP. Further, we observed the cleavage of PARP in the presence of THPP and light, where the full-length 116 kDa peptide cleaved into 89 kDa (Figure 2d). These results confirm that THPP

induces time-dependent apoptosis and that a majority of cells showed apoptotic cell death after 24 h. Further evidence was obtained through phase contrast microscopic morphological analysis of cells with and without irradiation (Supplementary Figure 7). We observed significant changes in the morphology from well attached and spread polygonal shaped cells to round and loosely attached ones when the cells were subjected to THPP plus light, confirming the apoptotic mediated cell death.

We employed DNA comet assay under neutral (non-denaturing) conditions to understand DNA damage and thereby to provide evidence for apoptotic mechanism. MDA MB 231 cells were treated with THPP and were incubated for 1 h. After 3 h post with and without PDT, the cells were subjected to agarose gel electrophoresis. Then, cells were stained with ethidium bromide and analyzed under fluorescence microscope. The fragmented nuclear DNA migrated in an electric field creating a comet-like appearance. The length of the tail is an indication of the extent of DNA damage. We observed the formation of DNA comets only when the cells were given THPP plus light. No such comets were observed with THPP in the dark, indicating thereby that THPP induces DNA fragmentation only upon irradiation (Supplementary Figure 8). In contrast, the usual photosensitizers like Photofrin act mostly on membranes through type II mechanism, whereas in DNA, singlet oxygen mainly induces oxidized guanine.³³

To further investigate the mechanism of cellular damage, we carried out internucleosomal DNA fragmentation studies through flow cytometric analysis.^{34,35} After 12 h of PDT treatment of IMR 32 cells with THPP, we observed 7.19% sub-G1 population, which interestingly increased to 39.83% after 24 h. This value is higher than that of 22.26% sub-G1 population observed in the case of Photofrin under similar conditions. These results confirm the enhanced apoptotic potential of THPP over Photofrin (Supplementary Figures 9 and 10). In the case of the apoptotic cells, the membrane phospholipid phosphatidylserine is known to be translocated from the inner to outer surface of the plasma membrane, thus exposing it to the external cellular environment. We employed Annexin V-

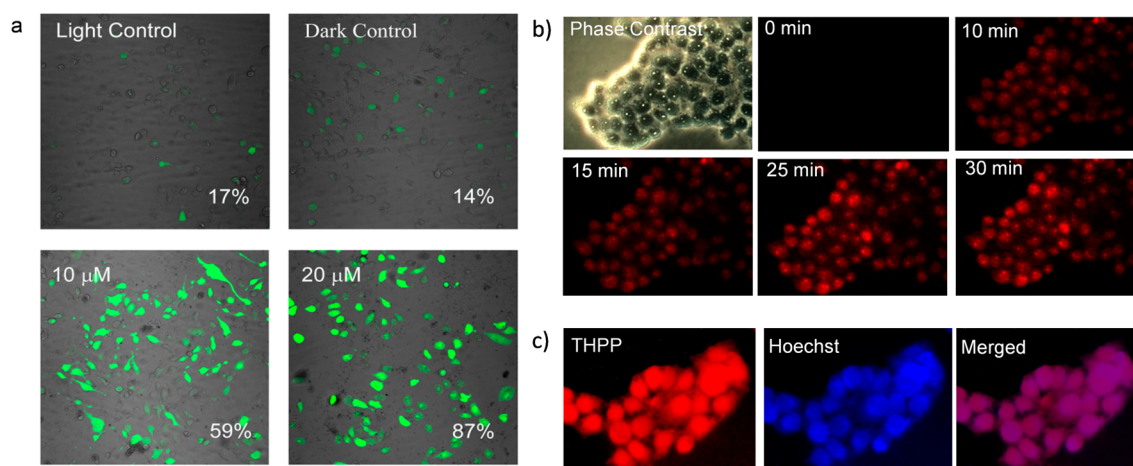


Figure 3. (a) Confocal microscopy images showing concentration-dependent green fluorescence in MDA MB 231 cells after PDT with THPP and incubation with ROS tracker CM-H₂DCFDA. (b) Time-dependent fluorescence changes showing cellular uptake of THPP in MDA MB 231 cells. (c) Fluorescence images showing localization of THPP in MDA MB 231 cells and perfect match of images with Hoechst dye.

FITC, which has high affinity for phosphatidylserine to quantify the formation of the apoptotic cells by fluorescence microscopy and flow cytometry. When IMR 32 cells were treated with THPP (10 μ M) in dark, we observed that most of the cells were negative for Annexin V-FITC, indicating the negligible formation of apoptotic cells. In contrast, the percentage of cells at the early apoptotic stage increased from 15.28% after 12 h to 69.93% after 24 h, when the cells were subjected to THPP and irradiation, thereby confirming that the mechanism of cell death is through the apoptotic pathway.^{36,37} Similar observations were made in the flow cytometric analysis of cells treated with THPP plus light (Figure 2e).

To understand the reactive oxygen species (ROS) involved, we quantified the generation of ROS in cells during the PDT treatment with THPP using 5-(and-6)-chloromethyl-2',7'-dichlorodihydrofluorescein diacetate acetyl ester (CM-H₂DCFDA) assay. In the reduced form, CM-H₂DCFDA is nonfluorescent, but after cellular oxidation followed by esterases mediated hydrolysis of acetate groups, results in a green fluorescent derivative. When MDA MB 231 cells treated with THPP (10 μ M) and irradiation followed by the addition of CM-H₂DCFDA, we observed fluorescence in 59% of cells, which increased to 87% with 20 μ M of THPP (Figure 3a). In contrast, light alone and THPP in the dark showed background values of 17% and 14%, respectively. These results demonstrate that only THPP upon excitation generates concentration-dependent ROS, which can be attributed to the oxidation of CM-H₂DCFDA and apoptosis mediated cellular death.

To investigate cellular uptake and localization of THPP, we employed MDA MB 231 cell lines and compared with Photofrin through fluorescence and confocal microscopic techniques. Interestingly, we observed fluorescence of THPP within 10 min (Figure 3b), whereas Photofrin showed negligible uptake even after 30 min of incubation (Supplementary Figure 11) under identical conditions. In the case of THPP, we observed that it localizes in the nucleus of the cell, showing intense red emission upon NIR excitation. The nuclear localization was confirmed by the counter staining method using the standard Hoechst dye (Figure 3c). The perfect match of the merged images of the MDA MB 231 cells with both THPP and Hoechst dye confirms the nuclear localization of THPP. The advantages of THPP as a live cell nucleus imaging

agent are its fast uptake, intense fluorescence, and NIR excitation, when compared to Hoechst dye, which requires UV light excitation.

In conclusion, we have designed a novel porphyrin derivative THPP that showed favorable photophysical properties, negligible dark toxicity, and high *in vitro* PDT activity³⁸ when compared to the clinically used sensitizer, Photofrin. Our investigations indicated that the biological activity of THPP involved predominantly ROS as the reactive intermediates and that the mechanism of cell death is due to apoptosis as confirmed through DNA fragmentation, PARP cleavage, comet assay, morphological changes, and flow cytometric analysis. Uniquely, THPP showed fast cellular uptake (<10 min) when compared to Photofrin and localized specifically in the nucleus of the cells. The advantages such as synthetic purity, ease of isolation, high water solubility, high triplet and singlet oxygen yields, fast cellular uptake, specific localization in the nucleus, and high *in vitro* photocytotoxicity make the newly synthesized porphyrin THPP an alternate sensitizer for PDT and live cell imaging applications.

METHODS

Calculation of Triplet Quantum Yields and Singlet Oxygen Quantum Yields. Quantum yields of triplet (Φ_T) and singlet oxygen generation (Δ^1O_2) of THPP and Zn-THPP were determined using an earlier described method^{25,26} (Supporting Information).

Determination of Cytotoxicity by Measuring Cellular Proliferation. 3-(4,5-Dimethyl-thiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) assay was used for determining cellular proliferation. To human cervical oral cancer cells (SCC 131) were added various sensitizers (2–20 μ M), incubated for 1 h, and irradiated at 25 °C using Waldmann PDT 1200 L system, fitted with 1200 W lamp (600–720 nm and 50 mW cm⁻²). After 24 h of incubation, 10 μ L of MTT (5 mg mL⁻¹ stock) was added. After 4 h, the supernatant was removed, 100 μ L of isopropyl alcohol was added, and the plates were covered with aluminum foil and kept on a shaker until formazan crystals were dissolved. The absorbance at 570 nm was measured, and percentage growth inhibition was calculated as % growth inhibition = (control – test)/control \times 100.

DNA Ladder Assay. SCC 131 cells with and without THPP and irradiation were harvested, washed with PBS, and incubated with lysis buffer at 37 °C for 2–3 h. DNA was extracted with isopropanol from the lysate. The precipitated DNA after centrifugation was air-dried and resuspended in tris-HCl overnight at 55 °C. One microgram of DNA was resolved through agarose gel electrophoresis. After staining with

ethidium bromide, the bands were visualized under the Gel Documentation System.

CM-H₂DCFDA Assay. MDA MB 231 cells were treated with sensitizers, incubated for 1 h, and irradiated for 30 min. Immediately after PDT, the cellular reactive oxygen content was determined using CM-H₂DCFDA probe according to the manufacturer's instructions (Invitrogen). The confocal images and differential interference contrast (DIC) images were acquired, merged, and processed using Nikon Imaging Software.

Cellular Uptake and Localization. MDA MB 231 cells were added with various sensitizers and incubated for different periods of time. The images were taken by using phase contrast and fluorescence microscopy. For confirming the nucleus localization, counter staining by standard Hoechst dye was used, and images were taken using appropriate filters.

■ ASSOCIATED CONTENT

■ Supporting Information

Experimental details including synthesis and supplementary Figures 1–11 showing photophysical and photobiological properties. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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Notes

The authors declare no competing financial interest.

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